

SYNTHESIS OF NEW (BENZIMIDAZOLYL)PIPERAZINES WITH AFFINITY FOR THE 5-HT_{1A} RECEPTOR VIA Pd(0) AMINATION OF BROMOBENZIMIDAZOLES

María L. López-Rodríguez, *† Alma Viso, Bellinda Benhamú, J. Luis Rominguera and Marta Murcia Departamento de Química Orgánica I, Facultad de Ciencias Químicas, Universidad Complutense, E-28040 Madrid, Spain

Received 25 May 1999; accepted 5 July 1999

Abstract: The synthesis of a new family of (benzimidazolyl)piperazines has been developed through Pd(0) mediated amination of 4- and 6-bromobenzimidazole derivatives. Preliminary studies showed that some of these compounds are potent 5-HT_{1A} receptor ligands. © 1999 Elsevier Science Ltd. All rights reserved.

The discovery of new ligands for the family of serotonin receptors is an area of active research in Medicinal Chemistry. Within this field the 5-HT_{1A} receptor has been intensively studied because it is involved in physiological processes and psychiatric disorders such as anxiety and depression. In addition to these therapeutic uses, serotonergic 5-HT_{1A} agonists have been recently suggested to be employed as neuroprotective agents² and the effect of these drugs may be therapeutically relevant. For these reasons we are currently involved in the design of new active and selective agents for this receptor. In this way, we have found that arylpiperazines 1a (Scheme 1) are potent 5-HT_{1A} ligands.³ 3D-QSAR studies⁴ showed that increasing the size of the substituent of the aromatic ring could enhance the affinity for the 5-HT_{1A} receptor and therefore we undertook the synthesis of arylpiperazines with bulkier aryl groups, 1b. One of the structures that we considered had a benzimidazole ring attached to the piperazine through carbon 4 or 5 and this required an efficient entry to 1-[benzimidazol-4(7)- and -5(6)-yl]piperazines.

Scheme 1

Benzimidazole derivatives are becoming increasingly important.⁵ Nevertheless, despite of their importance there are few methods available to prepare benzimidazole derivatives bearing substituents attached to the homocycle since most of the routes known are directed to functionalize carbon-2 and nitrogens (N-1, N-3).⁶ On the other hand, most methods to obtain arylpiperazines involve cyclization of aromatic amines and bis(haloethyl)amines.⁷ However, these procedures present some problems such as the use of toxic reagents and the difficult isolation of the products. Moreover, in our hands treatment of 4-amino-1-tritylbenzimidazole with bis(chloroethyl)amine under a number of different conditions did not yield the expected 1-(benzimidazol-4-yl)piperazine but complex crude mixtures with extensive decomposition. These disappointing results prompted us to explore alternative routes to reach our target compounds.

The metal-mediated coupling of aromatic halides and triflates with amines has become a powerful tool to synthesize aromatic amines, especially since Buchwald and Hartwig independently studied the process from a synthetic and mechanistic standpoint. Because of the broad scope of the amination reaction, we considered it suitable for our purposes. In this paper, we would like to report our results on the synthesis of 4(7)- and 5(6)-(benzimidazolyl)piperazines and on the synthesis of a new family of heteroarylpiperazines with activity for the 5-HT_{1A} receptor.

[†]Email: mluzlr@eucmax.sim.ucm

4(7)-Bromobenzimidazole 2¹⁰ was tritylated under standard conditions (TrCl/NaH/n-Bu₄NI/THF) yielding a single *anti* isomer, 4-bromo-1-tritylbenzimidazole, 3. Treatment of 3 with morpholine (1.4 equiv) at 85 °C in the presence of Pd₂dba₃·CHCl₃ (2-10%), dppp (2-10%), NaO'Bu (1.4 equiv) in toluene rendered 4-(benzimidazol-4-yl)morpholine 4a in 65% yield. We did not observe the corresponding reduction product (1-tritylbenzimidazole, 5)¹¹ in the ¹H NMR spectrum (300 MHz) of the crude reaction mixture (Scheme 2, Table 1, entry 1). Table 1 gathers our results on these couplings. Therefore, 1-methylpiperazine reacted with bromobenzimidazole 3 giving an excellent yield of 4b when Pd₂dba₃·CHCl₃ or Pd(OAc)₂ were used as catalyst along with dppp. However, when these conditions were applied to unprotected 2 no coupling product could be detected (Table 1, entries 2-4).

The behavior of 3 with an excess of piperazine (4 equiv) was also examined, and the corresponding (benzimidazolyl)piperazine 4c was produced under slightly different conditions. The use of Pd₂dba₃·CHCl₃/dppp did not yield the expected product but a mixture of starting material 3 and reduction product 5 (75:25) after 24 h of reaction, probably because of a competitive β-elimination process. Changing the catalyst system to Pd(OAc)₂/dppp gave a small amount of 4c in the crude mixture (25:50:25, 3:4c:5). However, the best result was obtained with Pd₂dba₃·CHCl₃/BINAP obtaining 4c with 87% yield (Table 1, entries 5-7). The corresponding dimer was not produced under these conditions; even in the presence of an excess of 3 and after prolonged reaction times (48 h), we only observed traces of a compound that was tentatively assigned as dimer along with 4c by ¹H NMR. Other amines submitted to the coupling reaction showed a lower reactivity. ¹² Thus, hexylamine (Table 1, entry 8) led to only 28% conversion after 48 h. In contrast, PhNH₂ afforded the corresponding amine 4e which was isolated pure in 54% yield at 60% conversion (Table 1, entry 9).

Scheme 2

Table 1. Pd(0) Amination of Bromobenzimidazoles

- 10 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1				
Entry	<u>Substrate</u>	RR'NH	Catalyst/Ligand	Product (yield%)a
1	3	morpholine	Pd2dba3 CHCl3/dppp	4a (65%)
2	3	1-methylpiperazine	Pd2dba3 CHCl3/dppp	4b (97%)
3	3	1-methylpiperazine	Pd(OAc) ₂ /dppp	4b (97%)
4	2	1-methylpiperazine	Pd(OAc) ₂ /dppp	_b ` ´
5	3	piperazine	Pd ₂ dba ₃ CHCl ₃ /dppp	_c
6	3	piperazine	Pd(OAc) ₂ /dppp	$4c^d$
7	3	piperazine	Pd2dba3 CHCl3/BINAP	4c (87%)
8	3	hexylamine	Pd2dba3 CHCl3/BINAP	4d ^e
9	3	aniline	Pd2dba3 CHCl3/BINAP	4e ^f (54%)
10	7a	piperazine	Pd2dba3 CHCl3/BINAP	_g ` ´
11	8	1-methylpiperazine	Pd ₂ dba ₃ CHCl ₃ /dppp	_g
12	7 b	piperazine	Pd(OAc) ₂ /BINAP	_h
13	7 b	piperazine	Pd(OAc) ₂ /BINAP	9 (60%) ⁱ

^aYield of isolated pure compounds. ^bNo reaction was observed. ^cA 75:25 mixture of 3 and 5 was observed in the reaction crude (¹H NMR, 300 MHz). ^dA 25:50:25 mixture of 3, 4c and 5 was observed in the reaction crude (¹H NMR, 300 MHz). ^eA 72:28 mixture of 3 and 4d was observed in the reaction crude (¹H NMR, 300 MHz). 4d was not isolated from this mixture. ^f60% conversion after 20 h of reaction. ^gA 50:50 mixture of starting material and 5 was observed in the reaction crude (¹H NMR, 300 MHz). ^hA 30:40:30 mixture of 7b, 5 and 9 was observed in the reaction crude (¹H NMR, 300 MHz). ^hA 15% of 5 was also obtained. Cs₂CO₃ instead of NaO'Bu was employed.

5- and 6-Bromobenzimidazoles¹³ were also considered as substrates for the amination reaction. Therefore, 6 was protected as before to produce a separable 50:50 mixture of the *N*-trityl derivatives 7a and 7b that in principle would converge to a single benzimidazole derivative after the coupling process and deprotection. However, unexpectedly we found that these regioisomers displayed a very different reactivity. Thus, 5-bromo-1-tritylbenzimidazole 7a was not a suitable substrate for the coupling reaction yielding starting material along with variable amounts of reduction product (5) under a number of conditions tried.¹⁴ The reaction with 5-iodo-1-tritylbenzimidazole 8¹⁵ was also tried and again starting material and reduction product 5 were obtained (Scheme 2, Table 1, entries 10-11). On the other hand, 6-bromo-1-tritylbenzimidazole 7b reacted more readily than its isomer 7a (entries 12 and 13). Thus, a low conversion to the corresponding (benzimidazolyl)piperazine 9 (30%) along with important amounts of reduction product 5 (40%) were produced after treatment of the bromo derivative 7b with an excess of piperazine, using Pd(AcO)₂/BINAP and NaO^tBu, however, the use of Cs₂CO₃ as base instead of NaO^tBu allowed to obtain the desired piperazine 9 along with small amounts of reduction product 5, that could be easily purified to render pure 9 in 60% yield.

To continue with our initial project on the design of new ligands for serotonin 5-HT_{1A} receptor we addressed the synthesis of heteroarylpiperazines 1b from the above (benzimidazolyl)piperazines 4c and 9. Therefore, 4c and 9 were treated with bromoderivative (\pm)-10^{3a} to yield (\pm)-11 and (\pm)-12. Standard acidic conditions¹⁶ (AcOH/H₂O/THF) allowed for the smooth removal of the protecting group on the final ligands to produce (\pm)-13 and (\pm)-14 in 85% and 65% yield respectively. Simple (benzimidazolyl)piperazines 4b and 4c were also submitted to this deprotection protocol rendering 15 and 16¹⁷ in good yields (Scheme 3).

a) NEt₃, CH₃CN, Δ , 80% for 11; 63% for 12. b) AcOH, THF, H₂O, 85% for 13, 65% for 14, 86% for 15, 66% for 16.

Preliminary studies on the in vitro affinity at 5-HT_{1A} receptors were carried out for 4b, 4c, 13, and 15 by radioligand binding assays, using [3 H]-8-OH-DPAT in rat cerebral cortex membranes. ¹⁸ Compounds 13 and 15 displayed high affinity for the 5-HT_{1A} receptor ($K_i = 1.21 \pm 0.02$ and 4.72 ± 0.50 nM, respectively), whereas tritylated derivatives 4b and 4c were inactive. These results indicate that arylpiperazines 13 and 15 are new potent 5-HT_{1A} receptor ligands. Thus, the fragment 1-(benzimidazol-4-yl)piperazine represents a novel and readily accessible pharmacophoric moiety in the research for new potent and selective 5-HT_{1A} ligands.

In summary, we have synthesized a new class of (benzimidazolyl)piperazines via Pd(0) amination of bromobenzimidazole derivatives. Some of these compounds have showed to be potent 5-HT_{1A} receptor ligands. The synthesis of new analogues, assessment for 5-HT_{1A} receptor affinity and further pharmacological characterization are in progress in our laboratory and the results will be published in due course.

Acknowledgment. This work was supported by DGICYT (Grant N° PB97-0282) and by CAM (Grant N° 08.5/0046/98). We also thank the UCM for a predoctoral grant to M. Murcia.

References and Notes.

- (a) Serotoninergic Neurons and 5-HT Receptors in the CNS; Baumgarten, H. G.; Göthert, M., Eds.; Springer-Verlag, Berlin, 1997.
 (b) Serotonin Receptors and their Ligands; Olivier, B.; van Wijngaarden, I.; Soudijn, W, Eds.; Elsevier, Amsterdam, 1997.
- (a) Semkova, I.; Wolz, P.; Krieglstein, J. Eur. J. Pharmacol. 1998, 359, 251. (b) Suchanek, B.; Struppeck, H.; Fahring, T. Eur. J. Pharmacol. 1998, 355, 95.
- (a) López-Rodríguez, M. L.; Rosado, M. L.; Benhamú, B.; Morcillo, M. J.; Sanz, A. M.; Orensanz, L.; Beneytez, M. E.; Fuentes, J. A.; Manzanares, J. J. Med. Chem. 1996, 39, 4439. (b) López-Rodríguez, M. L.; Morcillo, M. J.; Fernández, E.; Porras, E.; Murcia, M.; Sanz, A. M.; Orensanz, L. J. Med. Chem. 1997, 40, 2653. (c) López-Rodríguez, M. L; Morcillo, M. J.; Rovat, T. K.; Fernández, E.; Sanz, A. M.; Orensanz, L. Bioorg. Med. Chem. Lett. 1998, 8, 581. (d) López-Rodríguez, M. L; Morcillo, M. J.; Rovat, T. K.; Fernández, E.; Vicente, B.; Sanz, A. M.; Hernández, M.; Orensanz, L. J. Med. Chem. 1999, 42, 36. (e) López-Rodríguez, M. L.; Morcillo, M. J.; Fernández, E.; Rosado, M. L.; Orensanz, L.; Beneytez, M. E.; Manzanares, J.; Fuentes, J. A.; Schaper, K. J. Bioorg. Med. Chem. Lett., 1999, 9, 1679.
- López-Rodríguez, M. L.; Rosado, M. L.; Benhamú, B.; Morcillo, M. J.; Fernández, E.; Schaper, K. J. J. Med. Chem. 1997, 40, 1648.
- (a) Hansch, C.; Sammes, P. G.; Taylor, J. B. Comprehensive Medicinal Chemistry; Vol. 6, Pergamon Press, 1990.
 (b) Katz, B. A.; Clark, J. M.; Finer-Moore, J. S.; Jenkins, T. E.; Johnson, C. R.; Ross, M. J.; Luong, C.; Moore, W. R.; Stroud, R. M. Nature 1998, 391, 608. (c) Peng, Z.; Geise, H. J.; Zhou, X.; Peng, B.; Carleer, R.; Dommisse, R. Liebigs Ann./Recueil 1997, 27. (d) Cassidy, P. E. Thermally Stable Polymers Synthesis and Properties; Dekker, M., Ed.; N. Y., 1980.
- 6. For a recent review of the synthetic methods to obtain benzimidazoles, see: (a) Grimmett, M. R. Imidazole and Benzimidazole Synthesis; Meth-Cohn, O., Ed.; Academic Press, 1997. For recent papers on the synthesis of functionalized benzimidazole rings, see: (b) Edlin, C. D.; Parker, D. Tetrahedron Lett. 1998, 39, 2797. (c) Langer, P.; Döring, M. Synlett 1998, 399. (d) Meziane, M. A. A.; Rahmouni, M.; Bazureau, J. P.; Hamelin, J. Synthesis 1998, 967. (e) Konstantinova, L. S.; Rakitin, O. A.; Rees, C. W.; Sivadasan, S.; Torroba, T. Tetrahedron 1998, 54, 9639. (f) Mayer, J. P.; Lewis, G. S.; McGee, C.; Bankaitis-Davis, D. Tetrahedron Lett. 1998, 39, 6655. (g) Blettner, C. G.; König, W. A.; Rühter, G.; Stenzel, W.; Schotten, T. Synlett 1999, 307.
- (a) Chem. Abstr. 1960, 54, 9967b-i. (b) Chem. Abstr. 1961, 55, 8444h. (c) Mishani, E.; Dence, C. S.; McCarthy, T. J.; Welch, M. J. Tetrahedron Lett. 1996, 37, 319.
- For some representative papers in this field, see: (a) Guram, A. S.; Rennels, R. A.; Buchwald, S. L. Angew. Chem. Int. Ed. Engl. 1995, 34, 1348. (b) Sadighi, J. P.; Singer, R. A.; Buchwald, S. L. J. Am. Chem. Soc. 1998, 120, 4960.
 (c) Singer, R. A.; Buchwald, S. L. Tetrahedron Lett. 1999, 40, 1095. (d) Louie, J.; Hartwig, J. F. Tetrahedron Lett. 1995, 36, 3609. (e) Hamann, B. C.; Hartwig, J. F. J. Am. Chem. Soc. 1998, 120, 7369. (f) Ward, Y. D.; Farina, V. Tetrahedron Lett. 1996, 37, 6993. For recent reviews in this area, see: (g) Hartwig, J. F. Angew. Chem. Int. Ed. Engl. 1998, 37, 2046. (h) Frost, C. G.; Mendonça, P. J. Chem. Soc., Perkin Trans. 1 1998, 2615.
- Other arylpiperazines have been obtained by this method recently, see: (a) Morita, S.; Kitano, K.; Matsubara, J.; Ohtani, T.; Kawano, Y.; Otsubo, K.; Uchida, M. Tetrahedron 1998, 54, 4811. (b) Hong, Y.; Tanoury, G. J.; Wilkinson, H. S.; Bakale, R. P.; Wald, S. A.; Senanakaye, C. H. Tetrahedron Lett. 1997, 38, 5607. (c) Zhao, S.-H.; Miller, A. K.; Berger, J.; Flippin, L. A. Tetrahedron Lett. 1996, 37, 4463. (d) Nishiyama, M.; Yamamoto, T.; Koie, Y. Tetrahedron Lett. 1998, 39, 617.
- (a) Dandegaonker, S. H.; Revankar, G. R. J. Karnatak Univ. 1961, 6, 25; Chem. Abstr. 1965, 59, 10023e. (b)
 Tallec, A. Ann. Chim. 1968, 3, 164. (c) Montanari, F.; Passerini, R. Boll Fac. Chim. Ind. Bolog. 1953, 11, 42.
- 11. We prepared 1-tritylbenzimidazole 5 with comparative purposes by reaction of benzimidazole with NaH, TrCl, *n*-Bu₄NI in THF.
- 12. The coupling reaction with L-valine was also tried using CuI as co-catalyst or even in the absence of a palladium source, and unfortunately unreacted 3 was always recovered. (a) Ma, D.; Yao, J. Tetrahedron: Asymmetry 1996, 7, 3075. (b) Ma, D.; Zhang, Y.; Yao, J.; Wu, S.; Tao, F. J. Am. Chem. Soc. 1998, 120, 12459.
- 5(6)-bromobenzimidazole was obtained in a single step from benzimidazole, NBS and silica gel: Mistry, A. G.; Smith, K.; Bye, M. R. Tetrahedron Lett. 1986, 27, 1051.
- 14. Other conditions tried include the use of Pd₂dba₃/BINAP/Cs₂CO₃/Toluene, Pd(OAc)₂/BINAP/ NaO¹Bu/Toluene, and no positive results were found. Even when aniline was used to prevent β-elimination, we could not detect the corresponding coupling product upon examination of the crude reaction mixture.
- 15. 5(6)-iodobenzimidazole was obtained from o-nitroaniline, see: (a) Wilson, J. G.; Hunt, F. C. Aust. J. Chem. 1983, 36, 2317. (b) see 10c. Tritylation under standard conditions afforded a 50:50 mixture of 8 and its isomer.
- 16. Green, T. W.; Wuts, P. G. M. Protective Groups in Organic Synthesis; 2nd edition, J. Wiley & Sons, 1991.
- 17. UK Pat. Appl. GB 2,097,790. Chem. Abstr. 1983, 98, 107320e.
- 18. Clark, R. D.; Weinhardt, K. K.; Berger, J.; Fisher, L. E.; Brown, C. M.; MacKinnon, A. C.; Kilpatrick, A. T.; Spedding, M. J. Med. Chem. 1990, 33, 633.